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2 11. x^2y^3 x^3y^4 12. p p 6 3 a a 1 3 Express each using rational exponents. 13. x^5y^6
14. $27x^{10}y^5$ 15. $144x^6y^{10}$ 16. 21^3 c^7 17. $1024a^3$ 18. $36a^8b^5$ Express each using
radicals. 19. $64^{\frac{1}{3}}$ 20. $22a^3$ $2b^5$ 21. $s^{\frac{2}{3}}$ $t^{\frac{1}{3}}$ $v^{\frac{3}{2}}$ 22. $y^{\frac{3}{2}}$ 23. $x^{\frac{2}{5}}$ $y^{\frac{3}{5}}$
24. $(x^6y^3)^{\frac{1}{2}}$ $z^{\frac{2}{3}}$ $8^{\frac{2}{3}}$ $8^{\frac{1}{3}}$ Practice NAME _____ DATE _____ PERIOD _____
11-1

Practice Problems: Stoichiometry

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Chapter 11.3 Practice Problems - Drexel University

10-2 Practice Problems (continued) Date Class 1 If you burned 6.10×10^{24} molecules of C_2H_6 , produces 0.37 mol of ethane (C_2H_6), what mass of ethane did you burn? cut' sog 18 HOJ many formul units are in 5. N_2 gas. What volume will that gas occu b t mol C_2H_6 25. A canis r with a volum of 694 L contains how many moles of oxygen at / I mol 3R 10

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Honors Chemistry 11-3 practice problems Name: _____ See sample problem 5 on pg. 368 for the following problems: 1. Identify the limiting reactant when 5.1 g of lithium reacts with 1.5 L of fluorine gas at STP to produce lithium fluoride.

11 2 Practice Problems Continued Answers

2. Name the following compounds, determine the molar mass, determine how many O atoms are present in one molecule/formula unit, determine the grams of oxygen in 1.00 mole of the compound, and determine how many moles of O atoms in 8.35 grams of the compound. a) K_2CO_3 . b) $Ca(NO_3)_2$. c) Cu_2SO_4 . d) N_2O

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Chapter 11 Answers - Poudre School District

Practice Problems: Stoichiometry (Answer Key) Balance the following chemical reactions: a. $2 CO + O_2 \rightarrow 2 CO_2$ b. $2 KNO_3 \rightarrow 2 KNO_2 + O_2$ c. $2 O_3 \rightarrow 3 O_2$ d. $NH_4NO_3 \rightarrow N_2O + 2 H_2O$ e. $4 CH_3NH_2 + 9 O_2 \rightarrow 4 CO_2 + 10 H_2O + 2 N_2$ f. $Cr(OH)_3 + 3 HClO_4 \rightarrow Cr(ClO_4)_3 + 3 H_2O$ Write the balanced chemical equations of each reaction:

[Ms. Ose's Chemistry Website](#)

Honors Chemistry 11-2 Practice problems Name: _____ See sample problem 2 on pg. 357 for the following problems: 1. Determine the mass of zinc chloride produced when 5.30 grams of zinc react with hydrochloric acid. 11.0 g $ZnCl_2$ 2. Determine the mass of lithium hydroxide produced when 0.38g of lithium nitride reacts with water according the the following equation: $Li_3N + 3 H_2O \rightarrow ? NH_3$...

11 2 Practice Problems Continued Chemistry Answers

Chapter 11 Answers (continued) 36 Answers Algebra 1 Chapter 11 Chapter Project Activity 1: Writing 5280 ft; multiply 15 ft by ; ; Activity 2: Calculating A reasonable height would be less than 7 ft.

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Chapter 11.3 Practice Problems EXPECTED SKILLS: Know how to compute the dot product of two vectors. Be able to use the dot product to find the angle between two vectors; and, in particular, be able to determine if two vectors are orthogonal. Know how to compute the direction cosines of a vector.