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Name Date Class 3—3 Review and Reinforcement

16—3 Review and Reinforcement, pages 20-2J 5. endothermic 6. true 7. increases 8. true 9. The Concentrations of reactants and products, pressure, and temperature affect the equilibrium 10. Le Chatelier's principle states that if a change is imposed on a system at equilibrium, the equilibrium position will shift in the

[10-2 Review and Reinforcements - Front Page](#)

24—1 Review and Reinforcement (continued) 13. There are 3.29 g of iodine-126 remaining in a sample originally containing 26.3 g of iodine-126. The half-life of iodine-126 is 13 days. How old is the sample? 14. A sample that originally contained 2.5 g of rubidium-87 now contains 1.25 g. The half-life of rubidium-87 is 6×10^{10} years.

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10—2 Review and Reinforcement Mole Conversions Answer each of the following questions in the space provided. 1. How would you calculate the number of moles present in a given mass of a substance? 2. How would you calculate the number of particles present in a given number of moles of a substance? 23 10 3.

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14 terms. bigwill7714. 4-4 Review and Reinforcement. STUDY. PLAY. orbital. a region in space where an electron with a particular energy is likely to be found. ... Review and Reinforcement. 25 terms. A New Approach to the Atom + Electron Configurations. 58 terms. Chemistry Chapter 4: Arrangement of Electrons in Atoms. 44 terms.

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3—3 Review and Reinforcement (continued) Use the periodic table to determine how many protons, neutrons, and electrons are present in each of the following atoms. Write your answers in the spaces provided. Atom Protons Neutrons Electrons 12 iodine 125 _____ 13. potassium-39 / / 14. iron-56 _____ 3 Write the chemical symbol for each of the ions ...

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