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No. Equilibrium implies that the forward and reverse

reactions occur at the same rate. 15. The forward and

reverse reactions are both occurring, but with no net change

in concentration. -16—2 Review Reinforcement, pages 15-16

(H2J - [NH3JIHC1] 9. Keg - [Asd[C016 10. = ICO [C02]

12.

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set (13) What is the concentration of a solution

expressed as the number of moles of solute dissolved in

each liter of solution? molarity. What type of solution

contains as much solute as can possibly be dissolved

under existing conditions of temperature and pressure?

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2-3 Explore, page 15 : 1. Before mixing, the two

liquids filled each graduated cylinder completely.

After mixing, ... 2-5 Review and Reinforcement,

pages 28-29 . 1. heterogeneous 2. pure 3.

heterogeneous 4. pure 5. homogeneous 6.

heterogeneous 7. homogeneous 8. heterogeneous

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15. The amount of space that something occupies 17. ... 2

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