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ANSW

Apart by Leslie Cargile 1. What are atoms? a a. tiny particles that make up all matter b. tiny An Atom Apart particles that can only be seen with a microscope c. tiny particles that look like gnats d. particles that are so large they cannot be seen 2. What does the word A'tomos mean in ancient Greece? something that cannot parts; protons, be divided further 3. Parts Of The Atom Answer Key Worksheets - Kiddy Math This crossword puzzle, " An Atom Apart, " was created using the Crossword Hobbyist puzzle maker An Atom Apart R. Vaughan Science Grade force. Page 5/9 8 - ProProfs ... Splitting an atom is called nuclear fission, and the repeated splitting of atoms in fission is called a chain reaction. Nuclear fission is carried out in power plants in order to create energy. Scientists split atoms in order

to study atoms and the smaller parts they break into. This is not a process that can be carried out at home.

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Vocabulary Crossword. Across. 1. positively charged parts of an atom (protons) 6. negatively charged parts of an atom (electrons) 7. atoms are the building blocks for ... (molecules) 8. the number of electrons in atoms determine an. element's ____ properties (chemical) 9. neutrally charged parts of an atom (neutrons)

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Atoms are made up of three basic parts; protons, neutrons, and electrons. There is a core, or nucleus, and an electron cloud. The nucleus is made up of positively charged protons and neutral neutrons. The nucleus is held closely together by electromagnetic force. Protons and neutrons make up the nucleus of the atom. An Atom Apart <u>Crossword Answers -</u>

campaignbox.net An Atom Apart Answer Key Atoms are made up of three basic parts; protons, neutrons, and electrons. There is a core, or nucleus, and an electron cloud. The nucleus is made up of positively charged protons and neutral neutrons. The nucleus is held closely together by electromagnetic force. An Atom Apart - Super Teacher Worksheets 1. Name the three subatomic particles of an atom. Solution: An atom consists of three subatomic particles: Protons - positively

subatomic particles: Protons - positively charged; Electrons negatively charged; Neutrons - neutral in nature (no charge) 2. Helium atom has an atomic mass of 4 u and two protons in its nucleus. How many neutrons does it have? Solution:

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