

Answers For Chemistry Module 13

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[thermodynamics. Terms in this set \(29\) Enthalpy \(H\) The energy stored in an object. Change in Enthalpy \(H\) The energy change that accompanies a chemical reaction \(often referred to as just enthalpy\)](#)

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[in the problem or are in the Table 13.2, so we can apply](#)

[Equation \(13.9\) directly: \$\Delta H = \(12 \times 6 \times \(-242.6278 \text{ kJ}\)\)\$ \(2 X Using the rule of addition and subtraction, we can report our answer to the ones place, since the least precise bond energies have their last significant figure in the ones place. 3.](#)

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