
Answers Nivaldo Tro Chemistry

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Source(s): This was a real head scratcher, until I read in my chemistry book "Chemistry a Molecular Approach" by Nivaldo J. Tro, on page 257, the following: To get $\Delta H^\circ_{\text{rxn}}$ per mole of a particular reactant - a more general quantity - you simply divide by the number of moles that actually reacted, and this is what gave me the idea that maybe I should be dividing the obtained moles by 4 ...

A reaction mixture initially contains 2.9 ... - Yahoo Answers

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Consider the following reaction: $\text{CuS}(s) + \text{O}_2(g) \rightleftharpoons \text{Cu}(s) + \text{SO}_2(g)$ A reaction mixture initially contains 2.9 M O_2 . Determine the equilibrium concentration of O_2 if K_c for the reaction at this temperature is 1.5.