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The streptavidin solution was added to the washed preactivated beads at a ratio of 1:2 (v/v) and incubated at room temperature for 2 h with orbital agitation. The nonreacted groups on the beads were blocked by incubation with 1.5 mL of 0.1 M Tris-HCl, pH 8.5, for 1-2 h at room temperature with orbital agitation, followed by incubation with ...

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Inorganic fluorides. About 3 kg (6.5 lb) of metspar grade fluorite, added directly to the batch, are used for every metric ton of steel made. The fluoride ions from CaF<sub>2</sub> lower the melt's temperature and viscosity (make the liquid runnier). The calcium content has a tangential benefit in removing sulfur and phosphorus, but other additives such as lime are still needed.

**(PDF) Inorganic Chemistry Housecroft | Yurika Almada ...**

V. Barone and M. Cossi, "Quantum calculation of molecular energies and energy gradients in solution by a conductor solvent model," *J. Phys. Chem. A*, 102 (1998) 1995-2001. Barone98a: V. Barone, M. Cossi, and J. Tomasi, "Geometry optimization of molecular structures in solution by the polarizable continuum model," *J. Comp. Chem.*, 19 (1998) ...

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Atkins P, Overton T, Rourke J, Weller M, Armstrong F (2010). Shriver & Atkins' inorganic chemistry (5th ed. ed.). Oxford: Oxford University Press. ISBN 1429218207 Bache AD (1832), "An essay on chemical nomenclature, prefixed to the treatise on chemistry; by J. J. Berzelius", *American Journal of Science* vol. 22: pp. 248–277

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**Ionic compound - Wikipedia**

R.C. Ropp, in *Encyclopedia of the Alkaline Earth Compounds*, 2013 Calcium Chloride. Calcium chloride, CaCl<sub>2</sub>, is a typical ionic halide and is a solid at room temperature. Its molecular weight is 110.98 g/mol and its melting point is 772 °C. Very few natural minerals occur. The occurrence of a dihydrate (mineral "Sinjarite") and hexahydrate ("Antarcticite") is very rare and is connected

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