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answer each question in order to receive credit (1) Indicate whether each of the following is a permissible set of quantum numbers. If the set is not permissible, state why it is not a. $2, 4, 0, m=+1$ b. $n=5, l=3, m=3, n=4, l=3, m=2$ d. $n=0, l=1, m=0, m=4, l=1$ (2) Indicate whether each of the following is a permissible set of quantum numbers.

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Question 6. Define the following terms: (a) Atomic number (b) Mass number (c) Nucleons (d) Valence shell
Answer: (a) Atomic number: Atomic number refers to the number of protons present in an atom. It is denoted by Z .
Example: An atom of oxygen contains 8 protons. Therefore its atomic number is 8.

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3 7 3.

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1. In the nuclear model of the atom, protons (and neutrons) are housed in a small, dense nucleus. Electrons surround the nucleus in an area of mostly empty space. 2. If electrons are electrically attracted to nucleus and would, therefore, have centripetal acceleration in order to orbit the nucleus.

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