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program to answer the following questions
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answer key. Isotopes are atoms of the same
element with different numbers of neutrons
which results in different mass numbers.
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take the students about 30 - 45
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stru.

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Question 6. Define the following
terms: (a) Atomic number (b) Mass
number (c) Nucleons (d) Valence
shell Answer: (a) Atomic number:
Atomic number refers to the number
of protons present in an atom. It
is denoted by Z. Example: An atom
of oxygen contains 8 proton
Therefore its atomic number is 8.
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each question in order to receive
credit (1) Indicate whether each
of the following is a permissible
set of quantum numbers. If the set
is not permissible, state why it
is not a. 2,40m=+1 b. n.5,3, m-3
cn 4,1= 3, m2 d. n0,1-0, m4+1 (2)
Indicate whether each of the

following is a permissible set of quantum numbers.

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are electrically attracted to nucleus and would, therefore, have centripetal acceleration in order to orbit the nucleus.