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mass to calculate the average atomic mass for oxygen. Isotope information is provided below. Show all of your work and check your answer against the mass listed on the periodic table.

Isotope Natural Abundance on Earth (0/0)

Atomic Mass (amu) - 16.00 16.00 17.00 18.00

Average Atomic Mass - Financial

Chemistry

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15. Use one of the methods in Model 3 that gave the correct answer for average atomic mass to calculate the average atomic mass for oxygen. Isotope information is provided below. Show all of your work and check your answer against the mass listed on the periodic table. Isotope Natural Abundance on Earth (%) Atomic Mass (amu) 16O 99.76 15.9949

**Pogil Chemistry Average Atomic Mass Answer Key**

CP Chemistry 2015-16 Kaspriskie Average Atomic Mass POGIL Activity. Isotopes Review. Fill in the following table. Isotope

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Symbol Atomic Number Mass Number  
Protons Neutrons 1940 9 19 9 10 918 19 40  
19 21 S-31 16 31 16 15 Do all isotopes of an  
element have the same atomic number?

Explain your answer. YES.

### **Average Atomic Mass Pogil Answer- ebookdig.biz**

Calculate the average atomic mass. 6) Copper  
used in electric wires comes in two flavors  
(isotopes):  $^{63}\text{Cu}$  and  $^{65}\text{Cu}$ .  $^{63}\text{Cu}$  has an atomic  
mass of 62.9298 amu and an abundance of  
69.09%. The other isotope,  $^{65}\text{Cu}$ , has an  
abundance of 30.91%. The average atomic  
mass between these two isotopes is 63.546  
amu. Calculate the actual atomic mass of  $^{65}\text{Cu}$ .

### Pogil Chemistry Answer Key Average Atomic Mass

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Average Atomic Mass - Mr. Sault's Classroom. a.

Which isotope has an atomic mass closest to the average atomic mass listed on the periodic table?

24.mg b. Give a mathematical reason for your answer to part a. 24 Mg is the most common isotope and is thus most heavily "weighted" in the equation for average atomic mass. 17.