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Balancing Act Answer Key: Page 1 Problems 2 Ca + O2! 2 CaO N2 + 3 H2! 2 NH3 2 Cu2O + C ! 4 Cu + CO2 2 H2O2! 2 H2O + O2 Hint: Add the O atoms on the product side together when doing the counts. Page 2 Practice Problems 1. 2 Na + MgF2! 2 NaF + Mg 2. Mg + 2 HCl ! MgCl2 + H2 3. Cl2 + 2 Kl ! 2 KCl + I2 4. 2 NaCl!

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## Balancing Act Practice Name Balance each equation. Be sure ...

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### O NH O - Science Spot

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Balancing Equations Practice ANSWER KEY Part A: Identify the following parts of each chemical formula by circling the subscripts and drawing a square around the coefficients. H2 2 HCl 4 O 2 CH 4 3 CO 3 2 NaOH Part B: List the symbols for the atoms in each formula and give the number of each. C2H6 2MgO 4P 4O10 Balancing Act Worksheet Answer Key - worksheet

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Balancing Equations: Practice Problems

H = H = O = O = .1) Determine number of atoms for each element. 2) Pick an element that is not equal on both sides of the equation. 3) Add a coefficient in front of the formula with that element and adjust your counts.

Balancing Equations Practice Problems Worksheet Answers Key The download includes student worksheets, overhead key, and answer keys) Balancing Equations Practice (pdf) - A new version of Balancing Act that includes questions about subscripts and coefficients. An answer key is provided. Balancing Equations Practice - The Science Spot -MAFIADOC.COM

For your answer, type in the numbers that go in the blanks, with a comma and a space between each number (Ex. "2, 1, 3"). Terms in this set (44) 1, 1, 1 \_\_CaO + \_\_H?O ? \_\_Ca(OH)?

### **Balancing Act Practice Worksheet Answer Key**

Balancing Equations: Answers to Practice Problems 1. Balanced equations. (Coef?cients equal to one (1) do not need to be shown in your answers).

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### Name Atoms are not or during a chemical reaction. Mg + O

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