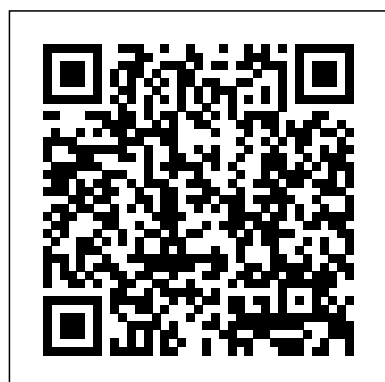


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Covalent Bonding and Shapes of. Chapter (c) C-Cl  $\hat{\text{I}} \text{ ' } + \hat{\text{I}} \text{ ' } \text{C-Cl}$

Chlorine is more electronegative than carbon Problem 1.6 Draw Lewis structures, showing all valence electrons, for these molecules.

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