
Calorimetry Gizmo Answers

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(ANSWER KEY)

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Gizmo Warm-up A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is

transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific heat capacity.

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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

Calorimetry Lab Gizmo : Lesson Info :

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Unit of measurement for heat and i learned the word from class in like the third grade Gizmo Warm-up A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up.

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Calorimeters can be used to find specific thermal abilities of a substance. You will use the Calorimetry Gizmo laboratory™ which determines the specific thermal abilities of various substances. 1. On the SIMULATION pane, select Bronze.

Calorimetry Gizmo Quiz Answers - HPD

Collaborative

Correct Answer: C. The final temperature of the lead-water system will be lower than the final temperature of the copper-water system. A blacksmith heats a 1,540 g iron horseshoe to a temperature of 1445°C before dropping it into 4,280 g of water at 23.1°C.

Student Exploration: Calorimetry Lab

Vocabulary: calorie ...

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Correct Answer: B 1 is a pain receptor, 2 is a temperature receptor, 3 is a light touch receptor,

and 4 is a strong pressure receptor [Book]

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Student Exploration- Calorimetry Lab (ANSWER KEY)

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Gizmo Warm-up A calorimeter is an insulated

container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific heat capacity.

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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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Calorimeters can be used to find a substance's specific heat capacity. You will use the

Calorimetry LabGizmo™ to determine the specific heat capacities of various substances.

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