

## Chapter 10 Chemical Quantities 247

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Chemical Quantities Chapter 10 Pg 247 Showing Work ...

10. Look at Figure 10.16 and Table 10.3. Name three compounds that have an empirical formula of CH. 11. Fill in the labels on the diagram below. MICROSCOPIC

INTERPRETATION MACROSCOPIC INTERPRETATION SO<sub>3</sub> molecule 1 mol SO<sub>3</sub> SO<sub>3</sub> composed of composed of S atom and sulfur atoms ( 1023) oxygen atoms 3 atoms and

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SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER (pages 287 – 296)

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