
Chapter 10 Chemical Quantities

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Work ...

10. Look at Figure
10.16 and Table
10.3. Name three
compounds that
have an empirical
formula of CH.

11. Fill in the labels
on the diagram
below.

**MICROSCOPIC
INTERPRETATI
ON**

**MACROSCOPIC
INTERPRETATI
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**SO₃ molecule
1 mol SO₃ SO₃**

composed of

**composed of S
atom and sulfur
atoms (1023)**

oxygen atoms 3

atoms and

CHAPTER

**10,Chemical Quan
tities(continued)**

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Quantities (Empiri

cal/Molecular

Formula) Chapter

7 - Chemical

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- A Fast Review!

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