

## Chapter 10 Chemical Quantities Test

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Honors Chemistry ... Chapter 10 - Chemical Quantities. 10.1 The Mole: A Measurement of Matter 10.2 Mole - Mass and Mole - Volume Relationships 10.3 Percent Composition and Chemical Formulas Chapter 10 Test Review 10.2 Practice Problems ANS. Chapter 11 - Chemical Reactions.

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CHAPTER 10: Chemical Quantities. CHAPTER 10: Chemical Quantities BASICS: • The basic unit that is used to determine the amount of a chemical substance is called a mole • A mole(mol) of a substance is equivalent to  $6.02 \times 10^{23}$  particles of that substance • The mole was founded by a scientist named Avagadro, and he decided to use the

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*Unit 4 - Chapters 3 & 10 - Mrs. Gingras' Chemistry Page*

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*Objectives Vocabulary Key Equations*

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CHAPTER 10: Chemical Quantities BASICS: • The basic unit that is used to determine the amount of a chemical substance is called a mole • A mole(mol) of a substance is equivalent to  $6.02 \times 10^{23}$  particles of that substance • The mole was founded by a scientist named Avagadro, and he decided to use the

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**Chapter 10 - Chemical Quantities - Preston Treend**

Chapter 10 (Chemical Quantities) Test Study Guide The mole is the SI unit used to measure the number of representative particles in a substance. A representative particle can be an atom, an ion, or a molecule, depending upon the way a substance commonly exists.

*Chapter 10 Chemical Quantities Chapter Test B Answers*

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Chapter 10 "Chemical Quantities" Vocab. Tools. Copy this to my account; E-mail to a friend ... Use these activities to learn the vocabulary presented in this chapter. A B; percent composition: a description of the relative amounts of each element in a compound: empirical formula: the lowest whole-number ratio of the atoms of the elements in a ...

### CHAPTER 10: Chemical Quantities

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Unit 4 discusses measurement in science as well as measurements that are specific to Chemistry. Chapter 3 deals with general math topics including significant digits and scientific notation. Chapter 10 introduces the idea of the mole and the use of equivalences to the mole as conversion factors.

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Chapter 10 Chemical Quantities 241 Section Review Objectives • Relate Avogadro's number to a mole of a substance • Calculate the mass of a mole of any substance • Describe methods of measuring the amount of something • Compare and contrast the atomic mass of an element and its molar mass Vocabulary Key Equations • moles representative particles • representative particles moles

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