## Chapter 10 Chemical Reactions Assessment Answers

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that a chemical ... List three types of physical evidence that indi-cate a chemical reaction has occurred. Answers may include release or absorption of energy, change in color, change in odor, formation ...

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Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n). 18. The decomposition of a substance by an electric current is called . 19.

chemical reactions. 20.

Chapter 7 TEST CHEMICAL REACTIONS

CHAPTER Date STUDY GUIDE Class Chemical Reactions Section 9.1 Reactions and Equations In your textbook, read about evidence of chemical reactions. For each statement, write yes if evidence of a chemieal reaction is present. Write no if there is no evidence of a chemical reaction, 2, 4, 5, 6, 8, A tomato smells rotten.

Assessment Chapter Test A

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A(n) orders the elements by the ease with which they undergo certain CHOICE. \_\_\_\_\_ 1. Which of these is not a sign of a chemical reaction? a. A gas is given off. c. Heat is released. b. The material dissolves. d. A color change occurs. \_\_\_\_\_ 2. The substance that is formed in a chemical reaction is called the ... chemical reaction is 19 g, what must the combined ...

Topic: Chapter 10: Types of Chemical Reactions (Test 1)

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Multiple choice test: Chemical Formulas and Equations TN SPI 807.9.8 -Interpret the results of an investigation to determine whether a physical or chemical change has occurred. TN SPI 807.9.11 - Recognize that in a chemical reaction the mass of the reactants is equal to the mass of the products (Law of Conservation of Mass)TN SPI 807.9.10 Identify the reactants and products of a chemical reaction.

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CHAPTER 10 SOLUTIONS MANUAL The MoleThe Mole Solutions Manual Chemistry: Matter and Change • Chapter 10 161 Section 10.1 Measuring Matter page 320 – 324 Practice Problems pages 323 – 324 1. Zinc (Zn) is used to form a corrosion-inhibiting surface on galvanized steel. Determine the number of Zn atoms in 2.50 mol of Zn. 2.50 mol Zn

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