
Chapter 10 Chemistry Vocab

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Chemistry

A B; radioactivity: the process in which an unstable atomic nucleus emits charged particles and energy:
radioisotope: any atom containing an unstable nucleus

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the SI unit representing 6.02×10^{23}
representative particles of a substance:
Avogadro's number: 6.02×10^{23} particles:
standard temperature and pressure (00 C, 1 atm)
the temperature and pressure at which one mole of gas occupies a volume of 22.4 L: molar

volume: volume of a gas that contains one mole of the gas, is 22.4 L at STP: Avogadro ...

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Avogadro ' s Law- a statement that the volume of a gas maintained at constant pressure is directly proportional to the number of moles of the gas. 10.

Ideal gas equation-an equation for the gases that embody Boyle ' s Law, Charles ' s Law, and Avogadro ' s Hypothesis in the form $PV=nRT$ 11.

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Chemistry 1405 Chapter 10 1. Stoichiometry – calculate the amount of reactants required for a chemical reaction to be complete, and the amount of products made after the reaction comes to an end. 2. Use of a chemical equation to obtain conversion factors – a conversion factor derived from

the coefficients of a balanced chemical equation interpreted in terms of moles.

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