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Chapter 11 – Thermochemistry – Heat and  
Chemical Change Chapter 11: 1 – 35, 57,  
60, 61, 71 Section 11.1 – The Flow of  
Energy - Heat Practice Problems 1. When  
435 J of heat is added to 3.4 g of olive oil at  
21? C, the temperature increases to 85? C.  
What is the specific heat of olive oil?  
Knowns:  $q = 435 \text{ J}$ ;  $m \text{ olive oil} = 3.4 \text{ g}$ ;  $?T = (85 \text{ ?C} - 21 \text{ ?C})$ ; Use  $C = q / m \text{ ??} T = 435$   
? ...

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11-Thermochemistry 12-Colligative Properties 13-Thermodynamics

14-Gas Phase, Solubility, Complex Ion Equilibria 15-Kinetics 16-Acid-Base Equilibria

17-Electrochemistry < Previous: Chapter Eleven listing is 2 pages: Next > Page 1: Page 2 : Chapter Eleven - Thermochemistry: Videos: Gummy Bear Oxidation This video shows the oxidation of a gummy bear with molten  $KClO_3$ . optional version ...

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coefficient. a whole number that  
appears before a formula in an  
equation. spectator ion. a  
particle not directly involved  
in a chemical reaction.  
combustion reaction. a reaction  
in which oxygen reacts with  
another substance, often

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producing light or heat.  
reactant . a starting substance  
in a chemical reaction. skeleton  
equation. a chemical ...

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Partial Pressure, Effusion



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