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Chapter 11 - Thermochemistry - Heat and Chemical Change Chapter 11: 1 - 35, 57, 60, 61, 71 Section 11.1 - The Flow of Energy - Heat Practice Problems 1. When 435 J of heat is added to 3.4 g of olive oil at 21°C, the temperature increases to 85°C. What is the specific heat of olive oil? Knowns:  $q = 435 \text{ J}$ ;  $m \text{ olive oil} = 3.4 \text{ g}$ ;  $\Delta T = (85^\circ\text{C} - 21^\circ\text{C})$ ; Use  $C = q / m \Delta T = 435 \text{ J} / (3.4 \text{ g} \times 64^\circ\text{C}) = 2.0 \text{ J/g}^\circ\text{C}$  ...