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2) Thermochemistry Review Lecture 11  
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Main elements of the kinetic theory of... Types of  
motion experienced by differen... What are  
perfectly elastic collisions?  
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Heat and Chemical Change  
Chapter 11: 1 - 35, 57, 60, 61,  
71 Section 11.1 - The Flow of  
Energy - Heat Practice Problems  
1. When 435 J of heat is added  
to 3.4 g of olive oil at 21°C,  
the temperature increases to  
85°C. What is the specific  
heat of olive oil? Knowns:  $q =$   
435 J;  $m$  olive oil = 3.4 g;  $\Delta T$   
= (85°C - 21°C); Use  $C = q / m$   
 $\Delta T = 435 \text{ J} / \dots$   
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*and ...*  
Thermochemistry Review Part  
1. Definition Questions: you  
should be able to use or  
apply the following terms. Be  
able to . write. complete  
definitions for, energy  
temperature standard state .  
potential energy heat

standard heat of formation .  
kinetic energy specific heat  
capacity exothermic. enthalpy  
endothermic Law of  
conservation of Energy.  
thermal Hess's Law isolated,  
closed, open system ...  
*Chapter 11 Thermochemistry*  
*Heat and Chemical Change*  
*Thermochemistry Equations*  
~~*Formulas*~~ ~~*Lecture*~~  
~~*Review*~~ ~~*Practice*~~  
~~*Problems*~~ Chem 30 Chapter 11 -  
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Law Chemistry Problems -  
Enthalpy Change - Constant  
Heat of Summation Enthalpy  
Change of Reaction  $\Delta H$   
Formation - Thermochemistry  
 $\Delta H$  Calorimetry Practice  
Problems Thermochemical  
Equations Practice Problems  
*Specific Heat Capacity*  
*Problems*  $\Delta H$  *Calculations*  
*- Chemistry Tutorial -*  
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Density, Molar Mass, Mole  
Fraction, Partial Pressure,  
Effusion **Tricks to solve  
Thermochemistry problems  
easily | Enthalpy of  
formation combustion Chapter  
5 - Thermochemistry: Part 11  
of 11 Ch. 11 Part 1:  
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11 11 Chapter 6  
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