

Chapter 11 Thermochemistry Review Answer Key

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Chapter 11 Thermochemistry Review Answer
Honors Chemistry Chapter 11 (Thermochemistry) Main elements of the kinetic theory of... Types of motion experienced by differen... What are perfectly elastic collisions?

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Chapter 11 - Thermochemistry - Heat and Chemical Change Chapter 11: 1 - 35, 57, 60, 61, 71 Section 11.1 - The Flow of Energy - Heat Practice Problems 1. When 435 J of heat is added to 3.4 g of olive oil at 21? C, the temperature increases to 85? C. What is the specific heat of olive oil? Knowns: $q = 435 \text{ J}$; $m \text{ olive oil} = 3.4 \text{ g}$; $\Delta T = (85 \text{ }^\circ\text{C} - 21 \text{ }^\circ\text{C})$; Use $C = q / m \Delta T = 435 \text{ ? ...}$

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. potential energy heat standard heat of formation
. kinetic energy specific heat capacity exothermic.
enthalpy endothermic Law of conservation of Energy.
thermal Hess's Law isolated, closed, open system
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formula in an equation. spectator ion. a particle
not directly involved in a chemical reaction.
combustion reaction. a reaction in which oxygen
reacts with another substance, often producing
light or heat. reactant . a starting substance in a
chemical reaction. skeleton equation. a chemical
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