

Chapter 12 Stoichiometry Work Answers

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[the question. ____ 1. The first step in most stoichiometry problems is to ____.](#) a. add the coefficients of

the reagents c. convert given quantities to volumes b. convert given quantities to moles d. convert given quantities to masses ____ 2.

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Quantitative calculations involving reactions in solution are carried out in the same manner as we discussed in Chapter 11. Instead of masses, however, we use volumes of solutions of known concentration to determine the number of moles of reactants. Whether we are dealing with volumes of solutions of reactants or masses of reactants, the coefficients in the balanced chemical equation tell us ...

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solvent, is produced by a reaction between methane and chlorine. ... 12 Stoichiometry Answers. ... Login to reply the answers Post; Still have questions? Get your answers by asking now. Ask Question + 100.

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Stoichiometry Problems. When we carry out a reaction in either an industrial setting or a laboratory, it is easier to work with masses of substances than with the numbers of molecules or moles. We will first present the method used in most other books for converting from the mass of any reactant or product to the mass of any other reactant or product using a balanced chemical equation outlined ...

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Acceptable answers include the idea of writing a ratio using the coefficients of two substances from a bal-