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stoichiometry of reactions in solution use
volumes of solutions of known concentration
instead of masses of reactants or products.
The coefficients in the balanced chemical
equation tell how many moles of reactants are
needed and how many moles of product can
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Calculations be solved in just four simple

steps: Stoichiometric Calculations:

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Stoichiometric Calculations are mostly

based on chemical formulas Formula

Mass: It is defined as the sum of the

atomic weights of each atom present in the

molecule of the substance. For example

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One can find the percentage of the mass of a compound that comes from each of the elements in the compound by using this equation: % element = (number of atoms)(atomic weight) (FW of the compound) x 100. Stoichiometry. Percent Composition.

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12.2 Stoichiometric calculations . As stated before, stoichiometry represents the quantitative relationships between amount of reactants used and amount of products formed. Remember that all stoichiometric calculations must begin with.... Balancing! An incorrectly balanced equation is going to give you incorrect predictions of products

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