

## Chapter 13 States Of Matter Test

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### Chapter 13: States of Matter - Chemistry by Anna

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all matter consists of tiny particles that are constantly in motion What are the three assumptions of the kinetic theory as it applies to gases? -The particles in a gas are considered to be small, hard spheres with an insignificant volume. -The motion of the particles in a gas are rapid, constant, and random.

States of Matter (chapter 13) Flashcards | Quizlet

Title: Chapter 13 States of Matter 1 Chapter 13 States

of Matter 2 Kinetic Theory as Applied to Gases Fundamental assumptions about gases. The particles in a gas are considered to be small, hard spheres with an insignificant volume. Between particles in a gas there is empty space. No attractive or repulsive forces exist between the particles. 3

Name Date Class STATES OF MATTER 13

Chapter 13 States of Matter137 SECTION 13.1 THE NATURE OF GASES (pages 385 – 389) This section introduces the kinetic theory and describes how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance. Kinetic Theory and a Model for Gases (pages 385 – 386) 1. Chemistry (12th Edition) Chapter 13 - States of Matter ...

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## Chapter 13: States of Matter

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### Chapter 13 States Of Matter

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### Chapter 13 States Of Matter Worksheet

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States of Matter - 13.1 The Nature of Gases - 13.1

Lesson Check - Page 424 8 including work step by step written by community members like you. Textbook

Authors: Wilbraham, ISBN-10: 0132525763, ISBN-13: 978-0-13252-576-3, Publisher: Prentice Hall

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18 States of MatterThree States of Matter - Solids,

Liquids And Gases | Science For Kids State Of Matter

Chemistry Class 11 | Chapter 5 Most Important Question

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There are three states of matter that we will learn about

in this chapter. (If you want to learn about more states

of matter, I can refer you to somebody.) Those three

states are solid, liquid, and gas. These three states

are quite different. The main difference is in their

particles.

Chapter 13: States of Matter Flashcards | Quizlet

Chapter 13 States Of Matter Chapter 13 States of

Matter137 SECTION 13.1 THE NATURE OF GASES

(pages 385 – 389) This section introduces the kinetic

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Chemistry & You; 13.1 The Nature of Gases - Sample

Problem 13.1; 13.1 The Nature of Gases - 13.1 Lesson

Check; 13.2 The Nature of Liquids - Chemistry & You; 13.2

The Nature of Liquids - 13.2 Lesson Check; 13.3 The Nature

of Solids - Chemistry & You; 13.3 The Nature of Solids -

13.3 Lesson Check; 13.4 Changes of State - Chemistry &

You