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Chapter 17: Thermochemistry

...
Thermochemistry is the scientific study of the heat

energy that is involved in or produced by chemical reactions and/or physical transformations. What can you tell us about how this topic is studied and what its uses are? Take the quiz and we'll find out together!

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thermochemistry. The study
of energy changes that occur
during chemical reactions and
changes in state. chemical
potential energy. The energy
stored in the chemical ...

Chapter 17

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Name _____ Date _____

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Thermochemistry Study
Guide 17.1 – 17.2

Thermochemical

Equations 1. Make the
following conversions: a.
444 cal to joules = 1.86

$\times 10^3$ J b. 1.8 kJ to joules
= 1.8×10^3 J c. 0.45 kJ
to calories = 1.1×10^2
cal 2. Classify each of
these processes as
endothermic or
exothermic: a. condensing
steam ...

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CHAPTER 17,

Thermochemistry

(continued) 2. What is

calorimetry? Calorimetry

is the accurate and

precise measurement of

heat change for chemical

and physical processes.

3. Use Figure 17.5 on page
511. Circle the letter next
to each sentence that is
true about calorimeters. G
The calorimeter container
is insulated to minimize
loss of heat to or

Chapter 17

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5 Lessons in Chapter 17:

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Chapter 17:

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SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505–510)
Study Guide Thermochemistry
 $Q = m C \Delta T$ $\Delta H = -q C = 4.184$
J moles $g \text{ } ^\circ\text{C}$ 1. When 150-g sample of KCl dissolves in 65.0 g of water in a calorimeter, the temperature

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Chapter 17
Thermochemistry183 ... 184
Guided Reading and Study Workbook CHAPTER 17, Thermochemistry (continued) 6. In thermochemical

calculations, is the direction of heat flow given from the ... thermochemical equation in the first paragraph on page 517 as a guide. SECTION 17.3 HEAT IN CHANGES OF STATE (pages 520–526)

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Chemistry Thermochemistry Study Guide Last document update: ago Chemistry Adv. includes vocab, thermochemistry, energy, heat capacity, specific heat,

enthalpy, and formulas. ... General Upper Level Chemistry Chapter 17 to 22. Verified and Explained Answers.

CHAPTER ...
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Thermochemistry KEY - Mr. Fischer
Ch 17 Thermochemistry Practice Test Matching Match each item with the correct statement below.
a. calorimeter d. enthalpy
b. calorie e. specific heat
c. joule f. heat capacity
_____ 1. quantity of heat

needed to raise the temperature of 1 g of water by 1°C ____ 2. SI unit of energy ____ 3. *Chemistry (12th Edition) Chapter 17 - Thermochemistry - 17 ...*

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Chemistry Chapter 17: Thermochemistry: Study Guide: STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Caleb_Tucker1. Terms in this set (55) Energy. The capacity for doing work or supplying heat. Thermochemistry. The study of energy changes that occur during chemical reactions and changes in state.

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Thermochemistry ...
CHAPTER 17,
Thermochemistry
(continued) GUIDED PRACTICE PROBLEM 12 (page 513) 12. When 50.0 mL of water containing 0.50 mol HCl at 22.5°C is mixed with 50.0 mL of water containing 0.50 mol NaOH at 22.5°C in a calorimeter, the temperature of the solution increased to 26.0°C .
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Authors: Wilbraham,
ISBN-10: 0132525763,
ISBN-13:
978-0-13252-576-3,
Publisher: Prentice Hall
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count as extra credit on your

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Enthalpy. Make the following
conversions: 444 cal to
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