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and Their Compounds;

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Chemistry of
Transition Metals;

19.3 Spectroscopic
and Magnetic
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pH only a small amount
even if large amounts of
acid or base are added. A
buffer system should
contain considerably
more acid than base.
Specific buffer systems

should be chosen based
on the pH that must be
maintained. 114 Study
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definition states that acids are

compounds that contain

hydrogen, and when placed in

aqueous solution, ionize to

form hydrogen (H^+ ions).

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