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The Arrhenius defintion states that acids are compounds that contain hydrogen, and when placed in aqueous solution , ionize to form hydrogen (H+ ions).
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[Ar]4 s1 3 d5; (d) Fe: [Ar]4 s2 3 d6; (e) Ru: [Kr]5 s2 4 d6. 3. (a) La: [Xe]6 s2 5 d1, La 3+: [Xe]; (b) Sm:
[Xe]6 s2 4 f6, Sm 3+: [Xe]4 f5; (c) Lu: [Xe]6 s2 4 f14 5 d1, Lu 3+: [Xe]4 f14. 5. Al is used because it is the
strongest reducing agent and the only option listed that can provide sufficient driving force to convert La
(III) into La.
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