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# Chapter 2 Basic Chemistry Packet Answers

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## Chemistry Comes Alive

Chapter 2 - Basic Chemistry 1. The Chemistry of LifeChapter 2 2. Chemistry All living things depend on chemistry The structure and function of all living things are governed by the laws of chemistry Why is it important to understand chemistry in biology? Biologists study chemistry because all living things are made of the same kinds of matter that make up nonliving

things

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Chapter 2 Basic Chemistry - Welcome to Coach Armstrong's class  
Section Review 2-2 1. a 2. b 3. c 4. c 5. a 6. b 7. c 8. Polarity in a water molecule is caused by an uneven distribution of electrons between the oxygen and hydrogen atoms. 9. The concentration of H ions determines whether a solution is acidic or basic. 10. Capillary action is the effect of water rising in a narrow tube against the force of ...  
**Biology Chapter 2: Basic**

## Chemistry | Chemistry Flashcards ...

Chapter 2 Objectives (IWBAT)  
The student evaluates the energy needs of the human body and processes through those needs 1. Differentiate matter from energy. 2. List four major energy forms, and provide one example of how each energy form is used in the body. 3. Define chemical element, and list the four elements that form the bulk of body matter.  
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Period: Date: ... pH 8-14 is basic or alkaline with 8,9 weak bases, 12,13,14 strong bases ...  
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1 . Human blood normally has a pH between 7.35 and 7.45. By comparison, saliva has a pH

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Review Packet Answer Key

2.1 ...

1 Chapter 2.1, 2.2 Review

Packet – Answer Key 2.1

The Nature of Matter Lesson Summary Atoms The atom is

the basic unit of matter, made up of three subatomic

particles. Protons have a positive charge and neutrons

carry no charge. Strong forces bind protons and neutrons

together in the nucleus.

[Chapter 2 - Basic Chemistry -](#)

Chapter 2: Chemistry Comes

Alive Basic Chemistry

Chapter 2: Chemistry Comes

Alive Matter The “ stuff ” of

the universe. Anything that

occupies space and has mass.

States of matter Solid Liquid

Gas Basic Chemistry Chapter

2: Chemistry Comes Alive

Energy Less tangible no mass,

does not take up space, & is

only measured by its effects

on matter.

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worksheet. Terms in this set (23)

Acids Bases and Salts. substances

that ionize in water; good

electrolytes. Base. Proton (H<sup>+</sup>)

acceptor. Salts and Buffers.

ionize in water to release

hydrogen ions and a negative

ion other than hydroxide (OH<sup>-</sup>)

Salts.

[Mrs. Niemoth / Chapter 2 Basic](#)

[Chemistry](#)

CHAPTER 2 Basic Chemistry.

BASIC CHEM REVIEW

Differentiate between matter and

energy. matter is anything that takes up space and has mass energy is the capacity to do work or put matter into motion

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Proteins are constructed from

building blocks called amino

acids; 20 common types of

amino acids are found in the

body. Amino acid sequence

determines the proteins

constructed (Figure 2.12).

Fibrous, or structural, proteins

are the basic structural materials

of the body.

Quia - Chapter 2: Basic

Chemistry

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Chapter 20 Lymphatic System;

Chapter 21 Immune System;

Chapter 23 Digestive System;

Chapter 24 Urinary System;

Chapter 26 Reproductive

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Development and Heredity;

Chapter 3 Cells; Chapter 4 Cell

Metabolism; Chapter 5

Microbiology Basics; Chapter 6

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## Tissues and Membranes