

Chapter 21 Chemistry

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When antimatter encounters ordinary matter, both are annihilated and their mass is converted into energy in the form of gamma rays (γ)—and other much smaller subnuclear particles, which are beyond the scope of this chapter—according to the mass-energy equivalence equation $E = mc^2$, seen in the preceding section. For example, when a positron and an electron collide, both are annihilated ...

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Chapter 21. Nuclear Chemistry. 21.5 Uses of Radioisotopes. Learning Objectives. By the end of this section, you will be able to: List common applications of radioactive isotopes; Radioactive isotopes have the same chemical properties as stable isotopes of the same element, but they emit radiation, which can be detected. If we replace one (or ...

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~~Aloud Chapter 21~~

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Organic Chemistry - McMurry - Chapter 21: Acyl Transfer

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[21.5 Uses of Radioisotopes - Chemistry](#)

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21: Nuclear Chemistry - Chemistry LibreTexts

The n:p ratio for Cr-53 is $29/24 = 1.21$; for Mn-55, it is $26/25 = 1.04$; for Fe-56, it is $33/26 = 1.27$.

Positron decay occurs when the n:p ratio is low. Positron decay occurs when the n:p ratio is low.

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