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A nuclear fuel. A fissionable isotope must be present in large enough quantities to sustain a controlled chain reaction. The radioactive isotope is contained in tubes called fuel rods. A moderator. A moderator slows neutrons produced by nuclear reactions so that they can be absorbed by the fuel and cause additional nuclear reactions. A coolant.

21.1 Nuclear Structure and Stability – Chemistry
Chemical Reactions A chemical reaction is a change in which one or more substances are converted into new substances Review: a substance is an element or a compound Chemical Reactions The reactants are the substances that exist before the reaction The products are the substances that exist after the reaction Conservation of Mass Conservation of Mass: The total mass of the products must always ...

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A balanced nuclear reaction equation indicates that there is a rearrangement during a nuclear reaction, but of subatomic particles rather than atoms. Nuclear reactions also follow conservation laws, and they are balanced in two ways: The sum of the mass numbers of the reactants equals the sum of the mass numbers of the products.

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Chapter 21 - Chemical Reactions. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. mspeckham. Terms in this set (41) chemical reaction. a change in which one or more substances are converted into new substances. reactants. substances that combine or change. products.

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Write correct formulas for the products in these decomposition reactions. 3) a- (k) Co_2 (g) CS_2 + CO_2 8) 9) g RbClO_3 10) (9 . Write correct formulas for the products in these single, replacement reactions. ... 21-ICI 4 SO_2 + 30 6) GO H No t) SO_2 (s) CCl_4 \rightarrow CCl_2 .

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Chapter 21: Coordination chemistry: reactions of complexes

3. reaction force 4. action force 5. The

force also will be 500 N because action-reaction forces are equal and opposite. 6. $p = m v = 2 \text{ kg } 10 \text{ m/s} = 20 \text{ kg} \cdot \text{m/s}$ 7. $p = m v = 2000 \text{ kg } 10 \text{ m/s} = 20,000 \text{ kg} \cdot \text{m/s}$ 8. the 2000-kg truck because it has a greater mass Chapter 4 1. energy 2. potential 3. kinetic 4. gravitational 5. speed Section 1 ...

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Study Guide and Reinforcement - Answer Key

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Chemical Reactions- Chapter 21. STUDY. PLAY. chemical reaction. is a change in which one or more substances are converted into new substances. reactants. the substances that react. products. the new substances that are produced. chemical equation. way to describe

chemical reaction using chemical formulas and other symbols.

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Chapter 21: Coordination chemistry: reactions of complexes. Student resources; Videos of chemical reactions. Video clips demonstrating some key chemical experiments Chapter 21. A collection of links to a range of useful websites. Lecturer resources; Figures, structures, and tables from the book.

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Nuclear chemistry is the study of reactions that involve changes in nuclear structure. The chapter on atoms, molecules, and ions introduced the basic idea of nuclear structure, that the nucleus of an atom is composed of protons and, with the exception of ${}^1_1\text{H}$, neutrons.

Answer Key Chapter 21 - Chemistry 2e | OpenStax

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