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Compounds (continued) SFCTION 22.4 HYDROCARBON Start studying RINGS (pages 709-711) This classify cyclic hydrocarbons. It also explains the bonding in benzene. Cyclic Hydrocarbons (page 709) 1. Chapter 22 Hydrocarbon <u>Compounds</u> StudyHippo.com The Hydrocarbon Compounds chapter of this Prentice Hall Chemistry **Companion Course helps** students learn the essential lessons associated with hydrocarbon compounds. PPT - Chapter 22 Hydrocarbon Compounds

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hexane: CH3CH2CH2CH2CH2CH3 38. a. propane b. octane c. pentane 39. 40. a. 2-methylbutane b. 2,3-dimethylbutane c. 3-methylhexane 41. The carbon-carbon bonds are non-polar and the carbonhydrogen Chapter 22 -Hydrocarbon **Compounds Flashcards** | Quizlet Hydrocarbon compounds Section 22.1: Hydrocarbons What is the relationship between carbon and

organic chemistry? 200 years ago scientists thought that only living organisms could make carbon compounds known in living cells Organic means related to or made by living things In 1828, A German Chemist, **Friedrich Wohler** Chemistry: Chapter 22: Hydrocarbon Compounds Flashcards Chapter 22 Hydrocarbon Compounds SECTION 22.1 HYDRO CARBONS(pages

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can use our background knowledge to know that since carbon has four valence electrons, it can form four covalent bonds Prentice Hall Chemistry Chapter 22: <u>Hydrocarbon</u> Compounds ... ch. 22 Hydrocarbon Compounds. Pearson Success Chapter 22. STUDY, PLAY. hydrocarbons. the simplest organic compounds containing only carbon and hydrogen. alkane. a hydrocarbon in which

there are only single covalent bonds. straight-UNSATURATED chain alkane, contain any number of carbon atoms, one after the other, in a chain; ethane difference between is simplest. Chapter 21: Hydrocarbons Start studying Chapter 22:

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