

Chapter 22 Hydrocarbon Compounds

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Chapter 22 Hydrocarbon Compounds 237. SECTION 22.1 HYDROCARBONS (pages 693 – 701)

This section describes the bonding in hydrocarbons and distinguishes straight-chain from branched-chain alkanes. It also provides rules for naming branch-chained alkanes. Organic Chemistry and Hydrocarbons (pages 693 – 694)

chapter 22: hydrocarbon compounds

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[Hydrocarbons | Saturated and Unsaturated Carbon Compounds Homologous series | Don't Memorise Naming Organic Compounds - IUPAC Nomenclature of Alkanes Organic Chemistry Nomenclature IUPAC Practice Review - Naming Alkanes, Alcohols, Alkenes \u0026 Alkynes Hydrocarbon Power!: Crash Course Chemistry #40](#)

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Substituted Hydrocarbons and Their Reactions substituted ...

? This hydrocarbon is called "2,3-pentadiene" because there are 2

(di) double bonds (ene) after 2 and 3 (2,3) of a 5 carbon line

(penta). It is named the same way for triple bonds, but with an

yne instead of ene. Next are isomers. Isomers are compounds with

the same molecular formula but different molecular structures.

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Chapter 22 "Hydrocarbon Compounds". Use these activities to learn the vocabulary and

major concepts presented in the chapter. molecules in which the atoms are joined in the

same order, but the arrangements of the atoms in space are different.

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This unit looks at aliphatic and aromatic hydrocarbons, including alkyl substituted hydrocarbons and

structural and geometric isomers. There is also a brief introduction to functional groups. Chapter 22 -

Hydrocarbon Compounds

Chapter 22 Hydrocarbon Compounds Flashcards

Chemistry Chapter 22 Hydrocarbons. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match.

Gravity. Created by. Wendy_Alexander3. Terms in this set (20) Organic Compound. Applied to all

carbon containing compounds with the primary exceptions of carbon oxides, carbides, and carbonates,

which are considered inorganic. Organic Chemistry is ...

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Chapter Test A A. Matching Match each term in Column 8 with (he cor-rrm description in

(jolumnA. Write the letter

ch 22 hydrocarbon compounds.pdf - n Chapter Test A A ...

an organic compound that contains only carbon and hydrogen. alkane. a hydrocarbon containing only single

covalent bonds; saturated hydrocarbons. straight-chain alkane. a saturated hydrocarbon that contains any

number of carbon atoms arranged one after another in a chain. homologous series.

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molecules of hydrocarbons, such as alkanes, are nonpolar molecules. Hydrocarbons. the simplest organic

compounds contain only carbon and hydrogen. Alkane. a hydrocarbon in which there are only single

covalent bonds. Straight-chain alkanes. contains any number of carbon atoms, one after the other, in a

chain.

[Chapter 22 Hydrocarbon Compounds](#)

1. Find the root word (ending in -ane) in the hydrocarbon name. Then write the longest carbon

chain to create the parent structure. 2. Number the carbons on this parent carbon chain. 3.

Identify the substituent groups. Attach the substituents to the numbered parent chain at the

proper positions. 4. Add hydrogens as needed.

[Chapter 22 Hydrocarbon Compounds | StudyHippo.com](#)

Hydrocarbon compounds Section 22.1: Hydrocarbons What is the relationship between carbon and organic

chemistry? 200 years ago scientists thought that only living organisms could make carbon compounds known in

living cells Organic means related to or made by living things In 1828, A German Chemist, Friedrich Wohler

[Chapter 22 Section 2 Hydrocarbons Part 2](#)

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[Authors: Wilbraham, ISBN-10: 0132525763, ISBN-13: 978-0-13252-576-3, Publisher: Prentice](#)

[Hall](#)

[SECTION 22.1 HYDROCARBONS \(pages 693 – 701\)](#)

Section 1: Hydrocarbons Hydrocarbons: Organic compound that contains only carbon and hydrogen. Alkane:

Hydrocarbon containing only single covalent bonds. Straight-chain: Saturated hydrocarbon that contains any

number of carbon atoms arranged one after the other in a chain. Homologous Series: Group of compounds in

which there is a constant increment of change in molecular structure from one ...

Chapter 22 hydrocarbon compounds Flashcards | Quizlet

Made up of about 80% methane, 10% ethane, 4% propane, and 2% butane. The remaining 4% consists

of nitrogen, helium, and hydrocarbons of higher molar mass. Propane and butane are separated from

the other gases in natural gas by liquefaction. Complete combustion of a hydrocarbon produces a blue

flame.

Chapters 22 & 23 - Organic Chemistry - Mrs. Gingras ...

Hydrocarbons are chemical compounds that contain the elements carbon and hydrogen only.

They are compounds that are obtained from the fossil fuel crude oil by a process called

fractional...

[Chapter 22: Hydrocarbon Compounds - Quizlet](#)

hydrocarbons results in a diverse group of organic compounds. CHAPTER 22 Substituted Hydrocarbons and

Their Reactions substituted Hydrocarbons and Their Reactions The larva of the Cerura vinula moth squirts

formic acid when threatened. Formic acid is a substituted hydrocarbon—a type of compound that forms when

one or more hydrogens

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Prentice Hall

Hydrocarbons - Hydrocarbons - National 4 Chemistry ...

Chapter 22 hydrocarbon compounds. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match.

Gravity. Created by. Liv20cederstrom. Terms in this set (23) Alkanes. a hydrocarbon containing only

single covalent bonds. Alkenes. Hydrocarbons with one or more carbon-carbon double bonds. alkyl

group. a hydrocarbon substituent.