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Because carbon has four valence electrons, car-bon atoms always form four covalent bonds. • The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches. • Molecules of hydrocarbons, such as alkanes, are nonpolar molecules. 22.2 Unsaturated Hydrocarbons Study Guide: Chapter 22 Hydrocarbon Compounds Flashcards ...

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Chapter 22: Substituted Hydrocarbons and Their Reactions. 784 Formic acid. BIGIdeaThe substitution of different functional groups for hydrogen atoms in hydrocarbons results in a diverse group of organic compounds.

22.1 Alkyl Halides and Aryl Halides. MAINIdeaA halogen atom can replace a hydrogen atom in some hydrocarbons.

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A controlled process by which hydrocarbons are broken down or rearranged into smaller, more useful molecules natural gas This is typically made up of 80% methane, 10% ethane, 4% propane, and 2% butane

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