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The First 10 Alkanes # of Carbons Name Formula H_{2n+2}

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marina_lynn8. Terms in this set (64) Atom or group of atoms that can take the place of a hydrogen in a parent hydrocarbon

molecule. substituent. Compounds that have the same molecular formula, but the atoms are joined in a different order.

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Modern Chemistry 181 Organic Chemistry CHAPTER 22

REVIEW Organic Chemistry SECTION 2 SHORT ANSWER

Answer the following questions in the space provided.

____ 1. ____ Hydrocarbons that contain only single covalent

bonds between carbon atoms are called (a) alkanes. (c)

alkynes. (b) alkenes. (d) unsaturated.

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Chapter 22: Amines. Summary. Amines (e.g. RNH_2) are

organic derivatives of ammonia, NH_3 (i.e. replace the N-

H with N-C bonds) and so they have certain similarities with

ammonia (e.g. basicity, nucleophilicity). Nitrogen containing

compounds are biologically very important (examples)

amines, amino acids, amides, proteins, vitamins etc.

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AP Chemistry Chapter 22 - Organic and Biological

Molecules 22.1 Alkanes: Saturated Hydrocarbons A.

Straight-chain Hydrocarbons 1. Straight-chain alkanes have the formula C_nH_{2n+2} 2. Carbons are sp^3 hybridized The First 10 Alkanes # of Carbons Name Formula (C_nH_{2n+2}) 1 Methane CH 4 2 Ethane C 2H 6 3 Propane C 3H 8 4