

# Chapter 6 Chemistry Answer Key

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CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds.

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Modern Chemistry 13 Chemical Bonding CHAPTER 6 STUDY GUIDE Chemical Bonding SECTION 5 MOLECULAR GEOMETRY SHORT ANSWER Answer the following questions in the space provided. 1. Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms.  
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Chapter 6 Chemistry in Biology 6.1 Formative Questions A B C 0% 0% 0% A. an equal number of protons and neutrons B. an equal number of protons and electrons C. an equal number of neutrons and electrons What causes the overall charge of an atom to be zero?

**Chemistry worksheet answers**  
1 . The spectrum consists of colored lines, at least one of which (probably the brightest) is red. 3 . 3.15 m 5 . 3.233 x x 10 ?19 J; 2.018 eV 7 .  
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18.6 Occurrence, Preparation, and Properties of Carbonates 18.7 Occurrence, Preparation, and Properties of Nitrogen 18.8 Occurrence, Preparation, and Properties of Phosphorus  
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