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A. an equal number of

protons and neutrons B. an equal number of protons and electrons C. an equal number of neutrons and electrons What causes the overall charge of an atom to be zero?
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CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions.

(c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds.
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SECTION 5 MOLECULAR GEOMETRY SHORT ANSWER Answer the following questions in the

space provided. 1. Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms.

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1 . The spectrum consists of colored lines, at least one of which (probably the brightest) is red. 3 . 3.15 m
5 . $3.233 \times 10^{-19} \text{ J}$;

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