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## Chapter 6 Covalent Bonding Answers

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*Chapter 8 Covalent Bonding Answers Key*  
Section 6.2 – Covalent Bonding. A  
covalent bond is a chemical bond in which  
two atoms share a pair of valence  
electrons. When two atoms share one pair  
of electrons, the bond is called a single  
bond.

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CHAPTER 6 REVIEW Chemical Bonding  
SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.  
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6 Chemical Bonding

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Section 6.2 Covalent Bonding Flashcards | Quizlet

Covalent Bond is a Is a chemical bond in which two atoms share valence electrons

Which ones show orbitals of atoms overlapping when a covalent bond forms? [ electrons dot, structural formula, space -filling, electron cloud]?

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short answer answer the following questions in the space provided 1 a a chemical bond between atoms results from the attraction between the valence electrons and of different atoms a nuclei c isotopes b inner electrons d lewis structures 2 b a covalent

## Chapter 6 – Chemical Bonds

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What does the line between two elements in a covalent bond represent? answer choices .

The process of electron gambling.

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