

# Chapter 6 Review Chemical Bonding Answer Key

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Modern Chemistry 41 Chemical Bonding  
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following questions in the space provided. 1. \_\_\_\_\_  
A chemical bond between atoms results from the  
attraction between the valence electrons and \_\_\_\_\_ of  
different atoms. (a) nuclei (c) isotopes (b) inner  
electrons (d) Lewis structures 2.

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an attraction between atoms that allows the formation of chemical substances that contain two or more atoms. The bond is caused by the electrostatic force of attraction between opposite charges, either between electrons and nuclei, or as the result of a dipole attraction.

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You â€™ need to be familiar with three types of chemical bonds for the SAT II Chemistry exam: ionic bonds, covalent bonds, and metallic bonds.  
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go.hrw.com/hrw.nd/gohrw\_rls1/pKeywordResult s?HC2%20CH06 Chapter 6: Chemical Bonding: Welcome to Modern Chemistry. Chapter 6 describes covalent bonds, ionic bonds, Lewis dot structures, resonance structures, and â€™ ;  
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**Lewis Dot Structures**  
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6 Chemical Bonding  
CHAPTER 6 REVIEW Chemical Bonding SECTION 5 SHORT ANSWER Answer the following questions in the space provided. 1. Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms. Pairs of valence electrons repel one

another. 2. In water, two hydrogen atoms are bonded to one oxygen atom. Why isn't water a linear molecule?

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the following questions in the space provided. 1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.

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Section 6.2 – Covalent Bonding A covalent bond is a chemical bond in which two atoms share a pair of valence electrons.

When two atoms share one pair of electrons, the bond is called a single bond.

Covalent vs Ionic Bond

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[ANSWER Answer the following](#)

questions in the space provided. 1. Use

the concept of potential energy to describe how a covalent bond forms

between two atoms. 2. Name two elements that form compounds that are exceptions to the octet rule. 3.

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Chemical bonding that results from the electrical attraction between positive ions and negative ions is called ionic bonding. In a purely ionic bond, the metal atom gives up its electron or electrons to the nonmetal atom. In covalent bonding, a bond forms from the sharing of electron pairs between two atoms. In a purely covalent bond, the