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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

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Chemistry Chapter 7 Review. Alloy. Anion. Cation. Chemical Bond. A mixture of elements that has metallic properties. A negatively charge ion. A positively charged ion. The force that holds two atoms together. Alloy. A mixture of elements that has metallic properties. Anion. A negatively charge ion. 13 Terms. DebraGibes.

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amu, the molar mass is x g/mol. False b.

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AP Chemistry Chapter 7 Review Questions Multiple-choice exercise. Use these values to assist you in answering some of the questions in this problems set: $h = 6.626 \times 10^{-34} \text{ J} \cdot \text{sec}$ $c = 2.9979 \times 10^8 \text{ m/sec}$. Show all questions $\leq = \geq$ An element having the configuration $[\text{Xe}]6s^1$ belongs to the Group: ?

