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(a) When two electrons are removed

from the valence shell, the Ca radius loses the outermost energy level and reverts to the lower $n = 3$ level, which is much smaller in radius. (b) The +2 charge on calcium pulls the oxygen much closer compared with K, thereby increasing the lattice energy relative to a less charged ion.

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Assessment. 3. The positive charge of a group A metal is equal to its group number. The charge of a group A nonmetal is the group number minus 8. Chapter 9

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9.27. Discuss briefly giving an example in each case the role of coordination compounds in : (a) biological systems, (b) analytical chemistry, (c) medicinal chemistry, and (d) extraction/metallurgy of metals. Ans: (i) Coordination compounds are of great importance in biological systems.

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