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PART I in the space provide fi ,  
write the letter oi the term or  
pin'ase that best completes each  
statement or best answers each  
question.1. Knowing the mole ratio  
of 21 reactant and product in a  
chemical reaction. would alio  
you to (leinnine a. the. energy  
released in the reaction.

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Writethe answers to the  
questions on the line to the  
left, and showyour work in  
the space provided. Use  
thetable belowto answer  
the questions. Element  
Symbol Atomic mass (amu)

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Bromine Br 79.90 Carbon C  
12.01 Chlorine Cl 35.45  
Hydrogen H 1.01 iodine I  
126.90 Iron Fe 55.85  
Element Symbol Atomic ...

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Multiple Choice Identify the  
choice that best completes  
the statement or answers

the question. \*\*\*\*You need  
to know all the rules for  
naming ionic, molecular and  
acidic compounds. \*\*\*\*You  
will have questions over ch  
1,2,4 and 25 as well. The  
majority of your test is  
naming though.

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you might not have even  
thought to ask! Take  
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a bulleted overview listing  
the topics that will be  
covered in the

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use the answers ... 3

contains 9 × 10<sup>23</sup> oxygen  
atoms. III. A 200 g sample  
of CaCO<sub>3</sub> contains 2  
moles of CaCO<sub>3</sub> a. I only  
b. II only c. III only d. I and  
III only e. I, II, and III

Practice Test Ch 3

Stoichiometry

Name \_\_\_\_\_ Per \_\_\_\_\_

$2\text{MnO}_2 + 4\text{KOH} + \text{O}_2 + \text{Cl}_2$   
 $? 2\text{KMnO}_4 + 2\text{KCl} + 2\text{H}_2\text{O}$

9. For the reaction  
above, there ...

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