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Chemistry Live! – Worked Solutions 76 Workbook Chapter 9 – The Mole Concept

W9.1 (a) 1 mole of Li atoms = 7 g (b) 1 mole of Na atoms = 23 g (c) 1 mole of Ca atoms = 40 g (d) 1 mole of Fe atoms = 56 g (e) 1 mole of Ag atoms = 108 g (f) 1 mole of Pb atoms = 207 g

W9.2 (a) 1 mole of Cl atoms =

35.5 g 10 moles of Cl atoms = 35.5×10

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