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**Stoichiometr**

**y Worksheet** KEY Balance  
**Chemfiesta** the  
**Answers** following  
Balancing equations:  
Equations 1) 1 N 2 + 3  
and Simple S F 2 2 NF 3  
toichiometry-2) 2 C 6 H

$10 \text{ H}_2\text{O} + 17 \text{ O}_2$   
 $12 \text{ CO}_2 + 10 \text{ H}_2\text{O}$   
 $3) \text{ HBr} + 1 \text{ KHCO}_3$   
 $1 \text{ H}_2\text{O} + 1 \text{ CO}_2$   
 $4) \text{ GaBr}_3 + 3 \text{ Na}_2\text{SO}_3$   
 $1 \text{ Ga}_2(\text{SO}_3)_3 + 6 \text{ NaBr}$   
 $5) \text{ SnO} + 2 \text{ NF}_3$   
 $3 \text{ SnF}_2 + 1 \text{ N}_2\text{O}$   
 Using the following  
**Stoichiometry Worksheet**  
**Chemfiesta Answers**  
 Balancing Equations and Simple Stoichiometry-KEY  
 Balance the following equations:  
 1)  $1 \text{ N}_2 + 3 \text{ F}_2 \rightarrow 2 \text{ NF}_3$   
 2)  $2 \text{ C}_6\text{H}_{10} + 17 \text{ O}_2 \rightarrow 12 \text{ CO}_2 +$

$10 \text{ H}_2\text{O} + 3) \text{ HBr} + 1 \text{ KHCO}_3$   
 $1 \text{ H}_2\text{O} + 1 \text{ CO}_2$   
 $4) \text{ GaBr}_3 + 3 \text{ Na}_2\text{SO}_3$  ... and the reagent that leads to this answer is the limiting reagent.  
**Both Balancing Equations and Simple Stoichiometry-KEY**  
**Balancing Equations Answer Key**  
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 The first step to balance the equation is to write down the chemical formula of reactants that are listed on the left side of the chemical equation. After this, you can list down the products on the right hand side of the chemical

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equation. There is an arrow between the sides, signaling the direction the reaction is happening in.

Chemfiesta Stoichiometry Practice Answers

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1)Using the following equation:  
 $2 \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$   
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1)Using the  
following equation:  
 $2 \text{NaOH} + \text{H}_2 \text{SO}_4 \rightarrow 2 \text{H}_2 \text{O} + \text{Na}_2 \text{SO}_4$

SO. 4 How many  
grams of sodium  
sulfate will be formed  
if you start with  
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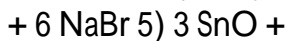
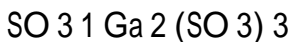
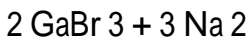
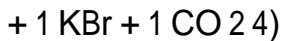
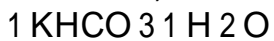
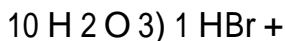
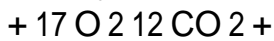
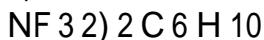
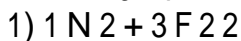
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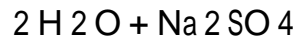
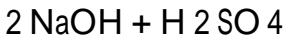
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Balance the

following equations:



following equation:



Stoichiometry

Homework Answer

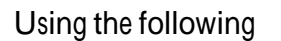
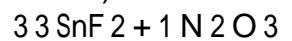
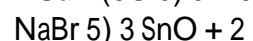
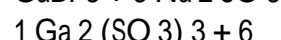
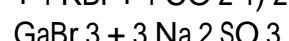
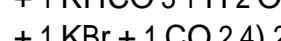
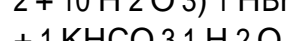
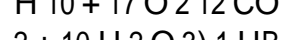
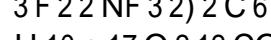
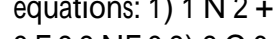
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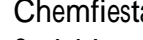
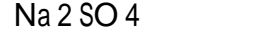
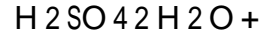
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Balance the following



Using the following



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Worksheet Answers

Balancing Equations

Answer Key

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Worksheet. 1) Using the

following equation: 2



How many grams of

sodium sulfate will be

formed if you start with

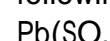
200 grams of sodium

hydroxide and you

have an excess of

sulfuric acid ( $\text{H}_2 \text{ SO}_4$ )? 2) Using the

following equation:



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