

# Chemfiesta Balancing Equations Simple Stoichiometry Answers

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The first step to balance the equation is to write down the chemical formula of reactants that are listed on the left side of the chemical equation. After this, you can list down the products on the right hand side of the chemical equation. There is an arrow between the sides, signaling the direction the reaction is happening in.  
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**Balancing Equations and Simple Stoichiometry-KEY**

Balancing Equations and Simple Stoichiometry-KEY Balance the following equations: 1)  $1 \text{N}_2 + 3 \text{F}_2 \rightarrow 2 \text{NF}_3$  2)  $2 \text{C}_6\text{H}_{10} + 17 \text{O}_2 \rightarrow 12 \text{CO}_2 + 10 \text{H}_2\text{O}$  3)  $1 \text{HBr} + 1 \text{KHCO}_3 \rightarrow 1 \text{H}_2\text{O} + 1 \text{KBr} + 1 \text{CO}_2$  4)  $2 \text{GaBr}_3 + 3 \text{Na}_2\text{SO}_3 \rightarrow 1 \text{Ga}_2(\text{SO}_3)_3 + 6 \text{NaBr}$  5)  $3 \text{SnO} + 2 \text{NF}_3 \rightarrow 3 \text{SnF}_2 + 1 \text{N}_2\text{O}_3$  Using the following equation:  $2 \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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Both Balancing Equations and Simple Stoichiometry-KEY

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Balancing Equations and Simple Stoichiometry-KEY Balance the following equations: 1)  $1 \text{N}_2 + 3 \text{F}_2 \rightarrow 2 \text{NF}_3$  2)  $2 \text{C}_6\text{H}_{10} + 17 \text{O}_2 \rightarrow 12 \text{CO}_2 + 10 \text{H}_2\text{O}$  3)  $1 \text{HBr} + 1 \text{KHCO}_3 \rightarrow 1 \text{H}_2\text{O} + 1 \text{KBr} + 1 \text{CO}_2$  4)  $2 \text{GaBr}_3 + 3 \text{Na}_2\text{SO}_3 \rightarrow 1 \text{Ga}_2(\text{SO}_3)_3 + 6 \text{NaBr}$  5)  $3 \text{SnO} + 2 \text{NF}_3 \rightarrow 3 \text{SnF}_2 + 1 \text{N}_2\text{O}_3$  Using the following

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