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 1.11 The force on a spring is described by: $F = Kx$ where K is the spring constant. First calculate K based on the earth measurement then g_{Mars} based on spring measurement on Mars.
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 $K_{Wact} = 125 \text{ mol}$ The solution is $T_f = 549.39\text{K}$ $c \text{ J}$

$\text{mol} ? K$ solution The actual is work 499.14K. is 25% Then greater h (b) Repeat the calculation with a temperature-dependent heat capacity?
 $10^2 T ? 3499 ? 10^5 T 2 + 7.464 ? 10^9 T 3 \dots CP(T) = 22.243 + 5977$ Assuming reversibility $T_f = 479.44\text{K}$.
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Al-Zaytoonah University of Jordan P.O.Box 130 Amman 11733 Jordan Telephone:

00962-6-4291511
00962-6-4291511 Fax:
00962-6-4291432. Email:
president@zuj.edu.jo. Student
Inquiries | ?????????? ??????:
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beaucag@uc.edu