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While chemical kinetics is concerned with the rate of a chemical reaction, thermodynamics determines the extent to which reactions occur. In a reversible reaction, chemical equilibrium is reached when the rates of the forward and reverse reactions are equal [the principle of dynamic equilibrium] and the concentrations of the reactants and Products no longer change.

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Textbook: Paul L. Houston, Chemical Kinetics and Reaction Dynamics,
Dover Publications. (Recommended)William L. Hase, Joseph Francisco,
Jeffrey I. Steinfeld, Chemical Reaction Dynamics, Prentice Hall.Raphael D.
Levine, Molecular Reaction Dynamics, Cambridge University Press.Tomas
Baer, William L. Hase, Unimolecular Reaction Dynamics: Theory and ...

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Book: An Introduction to Chemical Kinetics (Margaret Wright, 2004) -
get. Book chapter: Rates and Mechanism of Chemical Reactions
(Chapter 22 of Chemical Principles, 3rd Ed (1979) by Dickerson, Gray,
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Chemical kinetics - Wikipedia

Chemical change is guided and driven by energetics (thermodynamics), but the actual route it takes and the speed with which it occurs is the subject of "dynamics". Dynamics is itself divided into two general areas: kinetics, which deals with the rate of change and is the subject of this lesson.

Amazon.com: chemical kinetics and reaction dynamics

Chemical kinetics or reaction kinetic is the scientific study of the rates of chemical reactions.This includes the development of mathematical model to describe the rate of reaction and an analysis of the factors that affect reaction mechanisms.