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4.2 g Chapter 1 Section 1.4
Section 1.4 Questions (p. 26) 11.
1 , 2 , 3 , 3-, 2-, 1- Chapter 2
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The video below refers to Dynamic Equilibrium (under the section Equilibrium Systems) 7.1 p. 420 in the Nelson Textbook The video below is related to 7.2 Equilibrium Law and the Equilibrium Constant (p.429 in the

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Chapter 7: Chemical Equilibrium 7.1-4.

Section 7.1 Questions, page 428 1. (a) A sealed bottle of pop is a dynamic equilibrium because the rate of conversion between carbon dioxide and carbonic acid is equal in both directions in the closed system. (b) When I open the pop bottle, carbon dioxide leaves the bottle, so the equilibrium conditions no longer exist and a new equilibrium must form between the pop and the atmosphere.

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