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QUESTIONS Chemistry Review Unit Are You
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Chapter 7: Chemical Equilibrium 7.1-4.

Section 7.1 Questions, page 428 1. (a) A sealed

bottle of pop is a dynamic equilibrium because the rate of conversion between carbon dioxide and carbonic acid is equal in both directions in the closed system. (b) When I open the pop bottle, carbon dioxide leaves the bottle, so the equilibrium conditions no longer exist and a new equilibrium must form between the pop and the atmosphere.

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