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many moles of tungsten atoms are in 4.8×10^{25} atoms of tungsten? a. 8.0×10^2 moles c. 1.3×10^1 moles b. 8.0×10^1 moles d. 1.3×10^2 ...

Chemistry 2nd Semester Final Exam
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FINAL EXAM PRACTICE TEST *These are OLD TESTS. Your Final Exam will have questions VERY SIMILAR to your old tests! (* UNIT 7-GAS LAWS ____ 1. Which of the following statements best describes the movement of molecules in gases? ... CHEMISTRY 2ND SEMESTER ...

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Final Exam Practice Test (Semester 2) - Chemistry
2 Honors chemistry semester 2 final exam. Which of the following molecular properties is a main component of the chemical potential energy of matter? A. Vibrational motion C. Movement from place to place. B Honors chemistry semester 2 final exam. Intramolecular bonding D. Rotation about the molecules' centre of mass.

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Chemistry 2nd Semester Final Exam. STUDY. PLAY. valence electrons. Electrons on the outermost energy level of an atom. electron dot structures. diagrams that show valence electrons as dots. octet rule. Chemical compounds tend to form so that each atom, by gaining, losing, or sharing electrons, has an octet of electrons in its highest occupied ...

For each of the following questions,

please select the MOST correct answer
by clicking on its letter:

Chemistry Second Semester Final Exam
Study Guide

Honour Chemistry: Practice Final Exam
(Semester 2) Part A: Multiple Choice and
Numerical Response 1. Which of the following
equations is associated with the largest
energy change per mole of fluorine? A. $\text{F}_2(\text{g})$
 $\rightarrow \text{F}_2(\text{l})$ B. $19\text{ F} + 4\text{ He} \rightarrow 1\text{ O} + 22\text{ Na}$
C. $2\text{ F}_2(\text{g}) + 2\text{ H}_2\text{O}(\text{l}) \rightarrow \text{O}_2(\text{g}) + 4\text{ HF}(\text{aq})$

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