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...  
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Practice-Problem-6660020 book December 22, 2010 8:37 Chapter 16:  
Colligative Properties of Solutions 45 16-4. The mole fraction of  $(\text{NH}_4)_2\text{SO}_4(\text{aq})$  is given by  $x(\text{NH}_4)_2\text{SO}_4 = \frac{n(\text{NH}_4)_2\text{SO}_4}{n(\text{NH}_4)_2\text{SO}_4 + n\text{H}_2\text{O}}$  Because  $(\text{NH}_4)_2\text{SO}_4(\text{aq})$  is a strong electrolyte, it dissociates  
completely into  $\text{NH}_4^+(\text{aq})$  and  $\text{SO}_4^{2-}(\text{aq})$  ions. Assume a one ...  
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(18) Saturated solution. a solution containing the maximum amount of  
solute for a given amount of solvent at a constant temperature and  
pressure; an equilibrium exists between undissolved solute and ions ...

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1. A sample composed only of atoms having the same atomic number is classified as (1) a compound (3) an element (2) a solution (4) an isomer

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Chemistry/Chapter#16/Hydrocarbons/Lecture#02(2/2)/Nomenclature of  
Alkanes

16) HZ is a weak acid. An aqueous solution of HZ is prepared by  
dissolving 0.020 mol of HZ in sufficient water to yield 1.0 L of solution.

The pH of the solution was 4.93 at 25.0 ° C. The  $K_a$  of HZ is \_\_\_\_\_.

A)  $9.9 \times 10^{-2}$  ... Chapter 16. Practice Questions.tst Author: Satya

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Solutions Answers Chemistry Practice Chapter 16 Write if the

statement is true and if the statement is false. 1) According to the

Arrhenius theory, a neutralization reaction involves the combination  
of hydrogen ions and hydroxide ions to form water.